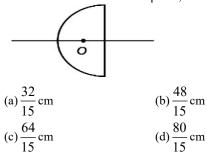


- 12. If force F is related with distance x and time t as $F = A\sqrt{x} + Bt^2$, the dimensions of $\frac{A}{B}$ is (a) M°L^{-1/2}T (b) ML^{-1/2}T⁻²
 - (a) $M^{\circ}L^{-1/2}T$ (b) $ML^{-1/2}T^{-2}$ (c) $M^{\circ}L^{-1/2}T^{2}$ (d) $M^{\circ}LT^{-2}$
- **13.** A verniercalliperhas 20 divisions on the vernier scale which coincide with 19 divisions on the main scale. The least count of the instrument is 0.1 mm. The length of one main scale division is
 - (a) 0.5 mm (c) 2 mm
 - (d)0.25 mm

(b) 1 mm

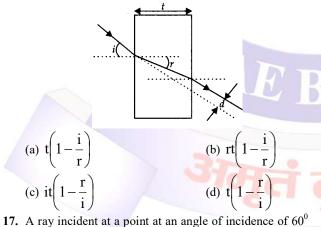
14. A glass hemisphere (μ =1.5) has a radius of curvature of 16 cm. A small object O is located on its axis halfway between the plane and spherical surface. The distance between two images, when viewed along the axis from the sides of the hemisphere, is



15. A convex lens forms inverted image of a real object on a fixed screen. The size of image is 9 cm. When lens is displaced 40 cm along principle axis it again forms a real image of size 4 cm on the screen. Focal length of the lens is .
(a)48 cm
(b) 100 cm

(a)48 cm	(b) 100 c
(c)30cm	(d)10cm
many of light is insident on	a think of

16. A ray of light is incident on a thick slab of glass of thickness t as shown in the figure. The emergent ray is parallel to the incident ray but displaced sideways by a distance d. If the angles are small then d is, :



enter a glass sphere of refractive index $\sqrt{3}$ and is reflected and refracted at the farther surface of the sphere. The angle between the reflected and refracted rays at this surface is : (a) 50⁰ (b) 60⁰ (c) 90⁰ (d) 40^{0} 18. Critical angle for light going form medium (i) to (ii) is θ . Thespedoflight in medium (i) is v, then then speed of light in medium (ii) is:

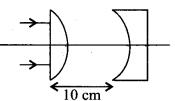
(a)
$$v(1 - \cos \theta)$$
 (b) $\frac{v}{\sin \theta}$

(c)
$$\frac{1}{\cos\theta}$$
 (d) $\frac{1}{(1-\sin\theta)}$

19. An air bubble in a glass sphere(μ=1.5) is situated at a distanced 3 cm form a convex surface of diameter 10 cm. At what distance from the surface will the bubble appear?

(a) 2.5 cm (b) -2.5 m (c) 5 m (d) -5 m

20. In the given figure. The radius of curvature of curved surface for both the plano-convex and plano- concave lens is 10 cm and refractive index for both is 1.5. The location of the final image after all the refractions through lenses is :



(a) 15 cm (b) 20 cm (c) 25 cm (d) 40 cm

- 21. The far point of a near sighted person is 6.0 m from her eyes, and she wears contacts that enable her to see distant objects clearly. A tree is 18.0 m away 2.0 m high. How high is the image formed by the contacts?
 (a) 0.1 m
 (b) 1.5 m
 (c) 0.75 m
 (d) 0.50 m
- **22.** A convex of lens of focal length 15 cm is placed on a plane mirror. An object is placed at 30 cm from the lens. The image is :
 - (a) Real, at 30 cm in front of the mirror
 - (b) Real, at 30 cm behind the mirror
 - (c) Real, at 10 cm in front of the mirror
 - (d) Virtual ,at 10 cm behind the mirror
- 23. Two identical glass $\mu_g = \frac{3}{2}$ equiconvex lenses of focal

length F are kept in contact. The space between the two

lenes is filled with water $\left(\mu_{w} = \frac{4}{3}\right)$. The focal length

of the combination is :

(a) f

(b)
$$\frac{f}{2}$$
 (c) $\frac{4f}{3}$ (d) $\frac{3f}{4}$

- 24. Two beams of red and violet color are made to pass separately through a prism (angle of the prism is 60°).In the position of minimum deviation, the angle of refraction will be:
 - (a) 30° for both the colors
 - (b) Greater for the violet color
 - (c) Greater for the red color
 - (d) Equal but not 30° for both the colors
- 25. For a glass prism ($\mu = \sqrt{3}$) the angle of minimum deviation is equal to the angle of the prism. The angle of the prism is : (a) 45^{0} (b) 30^{0} (c) 90^{0} (d) 60^{0}

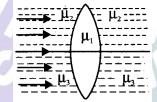
SEMRI KOTHI, SUPER MARKET, RAEBARELI MOBILE NUMBER 9792972355

- **26.** A ray of light is incident at 60° on one face of a prism of angle 30° and the emergent ray makes 30° with the incident ray. The refractive index of the prism is : (a) 1.732 (b) 1.414 (c) 1.5 (d) 1.33
- 27. The focal length of the lenses of an astronomical telescope are 50 cm and 5 cm. The length of the telescope when the image is formed at the least distance of distinct vision is :

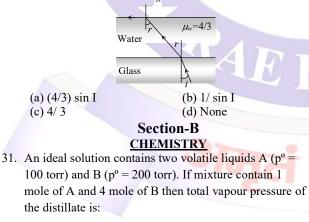
(a) 45 cm (b) 55 cm (c)
$$\frac{275}{6}$$
 cm (d) $\frac{325}{6}$ cm

28. A double convex lens, made of a material of refractive index μ_1 , is placed inside two liquids of refractive indices μ_2 is placed inside two liquids of refractive

indies μ_2 and μ_3 as shown. $\mu_2 > \mu_1 > \mu_3$ A wide, parallel beam of light is incident on the lens from the left. The lens will give rise to :



- (a) A single convergent beam
- (b) Two different convergent beams
- (c) Two different divergent beams
- (d) A convergent and a divergent beam.
- 29. In a compound microscope, the focal lengths of two lenses are 1.5cm 6.25 cm, If an object is placed at 2 cm from objective and the final image is formed at 25 cm from eye lens, the distance between the two lenses is (a) 6.00 cm (b) 7.75 cm (c) 9.25 cm (d) 11.0 cm
- 30. A ray of light is incident at the glass-water interface at an angle i, it emerges finally parallel to the surface of water, then the value of μ_g would be



(a) 150	(b) 180
(c) 188.88	(d) 198.88
1'	and DO in the

32. Two liquids A and B have P°_A and P°_B in the ratio of 1 :
3 and the ratio of number of moles of A and B in liquid phase are 1 : 3 then mole fraction of 'A' in vapour phase in equilibrium with the solution is equal to:
(a) 0.1
(b) 0.2

(c) 0.5

(a)

(c)

- (d) 1.0
- 33. Which of the following is less than zero for ideal solutions ?

(a) $\Box H_{mix}$	(b) $\Box V_{mix}$
(c) $\Box G_{mix}$	(d) $\Box S_{mix}$

34. One mole of a solute A is dissolved in a given volume of solvent. The association of the solute take place as follows: nA → A_n

If \Box is the degree of association of A, the van't Hoff factor i is expressed as:

$$i = 1 - \alpha \qquad (b) \quad i = 1 + \frac{\alpha}{n}$$
$$i = \frac{1 - \alpha + \frac{\alpha}{n}}{1} \quad (d) \quad i = 1$$

35. The degree of dissociation of an electrolyte is □ and its van't Hoff factor is i. The number of ions obtained by complete dissociation of 1 molecule of the electrolyte is:

(a)
$$\frac{i+\alpha-1}{\alpha}$$
 (b) $i-\alpha-1$
(c) $\frac{i-1}{\alpha}$ (d) $\frac{i+1+\alpha}{1-\alpha}$

36. If M_{normal} is the normal molecular mass and \Box is the degree of ionization of $K_3[Fe(CN)_6]$, then the abnormal molecular mass of the complex in the solution will be:

(a) $M_{normal} (1 + 2 \Box)^{-1}$ (b) $M_{normal} (1 + 3 \Box)^{-1}$

$$\begin{array}{ll} (a) \ K_2[PtCl_4] & (b) \ K_2[PtCl_6] \\ (c) \ K_3[PtCl_5] & (d) \ K[PtCl_3] \end{array}$$

38. A solute'S' undergoes a reversible trimerization when dissolved in a certain solvent. The boiling point elevation of its 0.1 molal solution was found to be identical to the boiling point elevation in case of a 0.08 molal solution of a solute which neither undergoes association nor dissociation. To what percent had the solute 'S' undergone trimerization?

A solution of x moles of sucrose in 100 grams of water freezes at □0.2°C. As ice separates the freezing point goes down to -0.25°C. How many grams of ice would have separated?

(a) 18 grams	(b) 20 grams
(c) 25 grams	(d) 23 grams

40. If □ is the degree of dissociation of Na₂SO₄, the vant Hoff's factor (i) used for calculating the molecular mass is :

 (a) $1 + \Box$ (b) $1 - \Box$

 (c) $1 + 2\Box$ (d) $1 - 2\Box$.

 41. The solubility of a gas in water depends on (a) Nature of the gas (b) Temperature (c) Pressure of the gas (d) All of the above 	(a) Normality (b) Molarity (c) Mole fraction (d) Mass percentage 53. The normality of $2.3 M H_2 SO_4$ solution is
42. Which is correct about Henry's law(a) The gas in contact with the liquid should behave	(a) $2.3 N$ (b) $4.6 N$ (c) $0.46 N$ (d) $0.23 N$ 54. The molarity of a solution made by mixing 50ml of
 as an ideal gas (b) There should not be any chemical interaction between the gas and liquid (c) The processor emplied should be high 	conc. H_2SO_4 (36N) with 50 ml of water is (a) 36 M (b) 18 M (c) 9 M (d) 6 M
 (c) The pressure applied should be high (d) All of these 43. The statement "If 0.003 moles of a gas are dissolved in 900 g of water under a pressure of 1 atmosphere, 0.006 	55. With increase of temperature, which of these changes(a) Molality(b) Weight fraction of solute
moles will be dissolved under a pressure of 2 atmospheres", illustrates(a) Dalton's law of partial pressure	(c) Fraction of solute present in water(d) Mole fraction56. Amorphous substances show
(b) Graham's law(c) Raoult's law(d) Henry's law	(A) Short and long range order(B) Short range order(C) Long range order
 44. The solution of sugar in water contains (a) Free atoms (b) Free ions (c) Free molecules (d) Free atom and 	 (D) Have no sharp M.P. (a) A and C are correct (b) B and C are correct (c) C and D are correct (d) B and D are correct
molecules 45. $25 ml$ of $3.0 M HNO_3$ are mixed with $75 ml$ of $4.0 M HNO_3$. If the volumes are additive, the molarity of the final mixture would be (a) $3.25 M$ (b) $4.0 M$	 57. The characteristic features of solids are (a) Definite shape (b) Definite size (c) Definite shape and size (d) Definite shape, size and rigidity
(a) $3.25 M$ (b) $4.0 M$ (c) $3.75 M$ (d) $3.50 M$ 46. The amount of anhydrous Na_2CO_3 present in 250 ml	58. Which one of the following is a good conductor of electricity(a) Diamond(b) Graphite
of $0.25 M$ solution is(a) $6.225 g$ (b) $66.25 g$ (c) $6.0 g$ (d) $6.625 g$	 (c) Silicon (d) Amorphous carbon 59. Diamond is an example of (a) Solid with hydrogen bonding
47. Dilute one litre 1 molar H_2SO_4 solution by 5 litre water, the normality of that solution is (a) $0.2N$ (b) $5N$	 (b) Electrovalent solid (c) Covalent solid (d) Glass
 (c) 10 N (d) 0.33 N 48. Which of the following has maximum number of molecules 	 60. Which is not a property of solids (a) Solids are always crystalline in nature (b) Solids have high density and low compressibility (c) The diffusion of solids is very slow
(a) 16 gm of O_2 (b) 16 gm of NO_2 (c) 7 gm of N_2 (d) 2 gm of H_2 49. 20 ml of HCl solution requires 19.85 ml of	(d) Solids have definite volume Section-C MATHS
0.01 M NaOH solution for complete neutralization. The molarity of <i>HCl</i> solution is (a) 0.0099 (b) 0.099	61. If $1 \le x \le \sqrt{2}$, then number of solutions of the equation $\tan^{-1}(x - 1) + \tan^{-1}x + \tan^{-1}(x + 1) = \tan^{-1} 3x$, is/are
 (c) 0.99 (d) 9.9 50. A mixture has 18g water and 414g ethanol. The mole fraction of water in mixture is (assume ideal behaviour 	(a) 0 (b) 1 (c) 2 (d) 3 62. If $\sin^{-1}x + \sin^{-1}y + \sin^{-1}z = \pi$, then $x^4 + y^4 + z^4$
of the mixture) (a) 0.1 (b) 0.4 (c) 0.7 (d) 0.9 51. The number of molecules in 4.25 g of ammonia is	$4x^2y^2z^2 = k (x^2y^2 + y^2z^2 + z^2x^2)$, where k is equal to -
approximately (a) 0.5×10^{23} (b) 1.5×10^{23}	(a) 1 (b) 2 (c) 4 (d) None of these 63. $\tan^{-1} x > \cot^{-1} x$ then find x- (a) [-1, 1] (b) $(-\infty, \infty)$ (c) $[1, \infty)$ (d) $(-\infty, 1]$
 (c) 3.5 × 10²³ (d) 2.5 × 10²³ 52. When the concentration is expressed as the number of moles of a solute per litre of solution it known as 	64. The value of

SEMRI KOTHI, SUPER MARKET, RAEBARELI MOBILE NUMBER 9792972355

 $\cos^{-1}\left(-\frac{1}{2}\right) - 2\sin^{-1}\left(\frac{1}{2}\right) + 3\cos^{-1}\left(-\frac{1}{\sqrt{2}}\right) - 4\tan^{-1}(-1)$ 72. The domain of the derivative of the function $f(x) = \begin{cases} \tan^{-1} x & , |x| \le 1\\ \frac{1}{2}(|x|-1) & , |x| > 1 \end{cases}$ is is equal to (b) $11\pi/4$ (c) $\pi/12$ (a) $7\pi/4$ (d) $25\pi/12$ 65. $\tan^{-1}\left(\frac{\pi}{4} + \frac{1}{2}\cos^{-1}\left(\frac{3a}{b}\right)\right)_{+} \tan^{-1}\left(\frac{\pi}{4} - \frac{1}{2}\cos^{-1}\left(\frac{3a}{b}\right)\right)_{is}$ (a) $R - \{0\}$ (c) $R - \{-1\}$ 73. Let $E = \{1, 2, 3, 4\}$ and $F = \{1, 2\}$. Then the equal to . number of onto functions from E to F is (a) $\frac{2b}{a}$ (b) $\frac{b}{2a}$ (c) $\frac{2}{3}\frac{b}{a}$ (d) $\frac{3}{2}\frac{a}{b}$ (a) 14 74. 'f ' is a real valued function not identically 66. $\cos^{-1}\left\{\frac{1}{2}x^2 + \sqrt{1-x^2}\sqrt{1-\frac{x^2}{4}}\right\} = \cos^{-1}\frac{x}{2} - \cos^{-1}\frac{x}{2}$ zero, satisfying f(x + y) + f(x - y) = 2f(x). f(y) $\forall x, y \in \mathbb{R}$. f(x) is definitely (a) Odd 1 x holds (c) Neither even nor odd (a) $|x| \le 1$ (b) $x \in R$ 75. Let α, β and γ be three positive real (c) $0 \le x \le 1$ (d) $-1 \le x \le 0$ numbers. Let $f(x) = \alpha x^5 + \beta x^3 + \gamma x, x \in \mathbb{R}$ 67. If $\sec^{-1}x = \csc^{-1}y$, then $\cos^{-1}\frac{1}{x} + \cos^{-1}\frac{1}{y} =$ and g: $R \rightarrow R$ be such that g(f(x)) = xfor all $x \in R$. If $a_1, a_2, a_3, \dots, a_n$ be in (a) π (b) $\frac{\pi}{4}$ (c) $\frac{-\pi}{2}$ (d) $\frac{\pi}{2}$ arithmetic progression with mean 68. If A &B are two sets such that n (A x zero, then the value of $f\left(g\left(\frac{1}{n}\sum_{i=1}^{n}f(a_{i})\right)\right)$ B) = 60 & n (a) = 12 also n (A \cap B) = K, then the sum of maximum & is equal to : minimum possible value of K is 1. (a)0 1. (a) 17 (b)12 2. (c) 9 2. (c)5 (d)7 The equation $x^2 - 4x + [x] + 3 =$ 76. 69. Let A and B be two sets. The set A x[x], where [x] denotes the greatest has 2016 more subsets than B. If $A \cap B$ integer function, has: has 3 members, then the number of 1. (a)Exactly two solutions in members in AUB is 1. (a) 10 (b) 11 2. (b) No Solution 2. (c)12 (d) 13 3. (c)A unique solution in 70. If in a class there are 200 students in which 120 take Mathematics, 90 4. (d)A unique solution in take Physics, 60 take Chemistry, 50 77. Let $f(x) = 2n^n + \lambda, \lambda \in \mathbb{R}, n \in \mathbb{N}$, and take Mathematics & Physics, 50 takeMathematics &Chemistry, 43 take Physics & Chemistry and 38 take Mathematics, Physics & Chemistry, (f(3) - f(2)) is then the number of students who 1. (a) 61 havetaken exactly one subject is 2. (c) 58 1. (a) 42 (b)56 2. (c) 270 (d) 98 71. If $f(x) = \cos[\pi^2]x + \cos[-\pi^2]x$, then common root if $\frac{d}{a}, \frac{e}{b}, \frac{f}{c}$ are in (a) $f\left(\frac{\pi}{4}\right) = 2$ (b) $f(-\pi) = 2$ (c) $f(\pi) = 1$ (d) $f\left(\frac{\pi}{2}\right) = -1$ value of λ will be-(a) 2

f(4) = 133, f(5) = 255. Then the sum of all the positive integer divisors of (b) 60 (d) 59 78. If a, b, c are in G.P. then the equations $ax^2 + 2bx + c = 0$ and $dx^2 + 2ex + f = 0$ have a (a) A.P. (b) G.P. (c) H.P. (d)None of these 79. If a non-zero root of the equation $x^2 + 2x + 3\lambda$ = 0 and $2x^2 + 3x + 5\lambda = 0$ is common, then the (c) - 1(d) 0

(b) $R - \{1\}$

(b) 16

 $(-\infty \infty)$

 $(-\infty, 1)$

 $(-\infty,\infty)$

(b) 1

(d) $R - \{-1, 1\}$

(c) 12

(b)3

(d) 27

(b) Even

(d)None of these

(d) 8

